## Safe Method of Use for Hazardous Substances of High Risk 1

## **Hydrofluoric Acid**

Hydrofluoric acid is a corrosive mineral acid that severely burn skin, eyes and mucous membrane. Vapours from anhydrous HF or its concentrated solutions can burn these tissues.

HF is similar to other acids in that the extent of the burn depends on the concentration, temperature and duration of contact.

HF differs however from other acids because the fluoride ion readily penetrates the skin causing destruction of deep tissue. Unlike other acids which are rapidly neutralised, this process may continue for days if left untreated

#### **WARNING:**

Burns with HF are usually very serious with potential for significant complications due to fluoride toxicity. Concentrated HF solutions or vapour may cause severe burns, tissue necrosis, metabolic imbalances, pulmonary oedema and life threatening cardiac arrythmias.

Burns from dilute solution, if left untreated, may progress to tissue necrosis. HF skin burns are accompanied by severe throbbing pain which is thought to be due the irritation of nerve endings by increase in extracellular potassium.

Any exposure to Hydrofluoric Acid MUST be medically evaluated.

#### A. Chemical Properties

- Hydrofluoric acid solutions are clear and colorless with a density similar to that of water. The most widely known property of HF is its ability to dissolve glass. It will also attack glazes, enamels, pottery, concrete, rubber, leather, many metals (especially cast iron) and organic compounds.
- Upon reaction with metals, explosive hydrogen gas may be formed.

#### B. Storage

- Use and store HF in polyethylene, polypropylene, and Teflon containers.
- Containers of concentrated Hydrofluoric acid must have prominent warning signs

#### **Toxicological Properties**

- Fluoride ions are both acutely and chronically toxic. Acute effects of HF exposure include extreme respiratory irritation, immediate and severe eye damage and pulmonary edema.
- Skin, eye, or lung exposure to concentrated (>50%) HF solutions will cause immediate, severe, penetrating burns.
- Exposure to less concentrated solutions may have equally serious effects, but the appearance of symptoms can be delayed for up to 24 hours.
- If you are exposed to hydrofluoric acid SEEK MEDICAL ATTENTION IMMEDIATELY.

#### **Personal Protective Equipment**

- The ACGIH ceiling limit and OSHA TWA for HF is 3 PPM.
- Fume Hoods *shall* always be used when working with HF. The sash of the fume hood *shall* be as low as reasonably practical
- Eye protection in the form of safety glasses or goggles and a face shield *shall* be used.
- Neoprene, viton/neoprene or butyl neoprene gloves *shall* be worn. (Hydrofluoric Acid burns around the fingernails are extremely painful, difficult to treat, and may require surgical removal of the nail).
- Barrier cream and nitrile gloves may be worn where short term use is anticipated.
- It is also recommended that an acid resistant apron be used (Some clothing is able to absorb the toxic material and maintain it close to the skin).
- In order to warn and protect others from the hazard of HF, a warning sign indicating the use of HF *shall* be posted.

#### Using Hydrofluoric Acid Safely

- *Never* use Hydrofluoric Acid when working alone or after hours.
- Hydrofluoric Acid may be used when working alone during normal working hours provided the Lab Manager have been alerted and at least one experienced person is in the general vicinity.
- All lab personnel, not just those who will be using Hydrofluoric Acid, should be informed of the dangers of this chemical and the emergency procedures necessary in case of an accident. A sign should be posted to alert people that work with Hydrofluoric Acid is in progress.
- All persons who will be using Hydrofluoric Acid must be made aware of its properties and trained in proper procedures for use and disposal.
- Laboratories which keep or use Hydrofluoric Acid gas or concentrated solutions (>1% Hydrofluoric Acid) should have these emergency procedures on hand as well as an SDS.

- Laboratories which keep or use Hydrofluoric Acid gas or concentrated solutions (>1% Hydrofluoric Acid) must have an operational safety shower and eye wash in their laboratory.
- Before beginning any procedure involving Hydrofluoric Acid, make sure the access to the emergency shower and eyewash is unobstructed.
- Only experienced persons specifically designated by the Lab Manager who are familiar with its properties shall handle the concentrated acid.
- All procedures involving HF shall be conducted in a certified fume hood.
- Before beginning work involving HF the following items shall be readily on hand:
  - 1. Container of calcium gluconate gel. This gel must be inspected before each use of HF or at least monthly to ensure the gel has not been removed or has not reached the expiration date. If a tube of the gel has been opened, a new container must be purchased and the old container discarded. No work with HF can be done with an expired tube of calcium gluconate gel.
  - 2. A small supply of calcium carbonate or calcium hydroxide for spills shall also be kept near the fume hood where the work will be conducted.
  - 3. A spare set of HF acid resistant gloves
  - 4. Copy of these procedures and MSDS to take to the emergency room.

#### Working with Hydrofluoric Acid or concentrated HF solutions (> 1%):

- Work in a fume hood with the sash as low as possible. Wear goggles and a face shield. Wear a long-sleeved, buttoned lab coat, pants or long skirt, and closed-toe shoes.
- Wear Neoprene or other Hydrofluoric Acid resistant gloves (Hydrofluoric Acid burns around the fingernails are extremely painful, difficult to treat, and may require surgical removal of the nail). A chemically resistant apron is also recommended.

#### **Spills**

- If a small quantity (100 ml or less) of dilute Hydrofluoric Acid solution is spilled, clean it up by applying powdered calcium carbonate or calcium hydroxide, or use a commercial Hydrofluoric Acid spill kit.
- If a larger amount is spilled, or the acid is concentrated, contain the spill as best you can, evacuate the area, and call 85000. Avoid exposure to the vapours.

# Emergency Response Procedures: Skin Exposure:

- Move the victim immediately under an emergency shower or other water source and flush the affected area with large amounts of cool running water for at least 1 minute. While the victim is flushing with water, they should also take off all clothing, shoes and jewellery, removing goggles last. Close your eyes, face water flow and pull goggles over head. BE EXTREMELY CAREFUL NOT TO CONTAMINATE YOURSELF (USE Neoprene gloves).
- Remove all contaminated clothing while flushing with water. Some clothing is able to absorb the toxic material and maintain it close to the skin.
- While the victim is being rinsed with water, someone should call to arrange treatment by medical personnel. Call Unisafe 85000 or Emergency Services (111) and tell the dispatcher that a person that has been in exposed to Hydrofluoric Acid and the location. The hospital may need to be alerted.
- Immediately washing off the acid is of primary importance!
- After the affected area is flushed with copious amounts of water for at least one minute, 2.5% calicum gluconate gel is to be applied using these guidelines.
- In order to prevent cross contamination, the victim should self apply the calcium gluconate gel.
- If the victim is unable to self apply, anyone present can apply the gel after putting on the neoprene gloves.
- Do not use latex gloves; they are not effective against HF. Note the time when the calcium gluconate gel was first applied to the contaminated site. Provide this information to the Emergency Services team.
- The victim must be escorted to the hospital by the responding person or assisting lab personnel.
- A copy of the SDS and these emergency procedures must be also taken to the hospital.

### **Eye Exposure:**

- Immediately flush eyes for at least 5 minutes with copious cool flowing water.
- EMERGENCY SERVICES SHOULD BE NOTIFIED IMMEDIATELY. Tell the dispatcher that a person that has been in exposed to Hydrofluoric Acid and the location. The hospital may need to be alerted.

• If a sterile 1% calcium gluconate solution is available and MEDICAL PERSONNEL are available then following the 5 minutes wash time, 1% calcium gluconate irrigation should be started.